

### admium may be present in natural waters as hydrated ions, chloride salts, complexed with inorganic ligands, or chelated to form complexes with organic ligands. Soils may contain various components, such as mixed hydroxides, oxides, silicates, and sulphides, that can complex with cadmium and influence its fate in water. Adsorption and ion exchange can occur with clay, silica, or organic matter, as cadmium has a high affinity for negatively charged particle surfaces. Consequently, cadmium tends to be removed rapidly from solution and accumulate in soils or sediments. Changes in environmental conditions, however, such as reduced pH, changes in redox status (e.g., due to spring and fall turnover), and biological and chemical oxidation of organic matter, have the potential to remobilize and transport cadmium to other compartments of the ecosystem.

For more information on the use, environmental concentrations, and chemical properties of cadmium, see the fact sheet on cadmium in Chapter 4 of *Canadian Environmental Quality Guidelines*.

## Water Quality Guideline Derivation

The Canadian water quality guidelines for cadmium for the protection of agricultural water uses were developed based on the CCME protocol (CCME 1993).

## **Irrigation Water**

A water quality guideline for cadmium of  $5.1 \ \mu g \cdot L^{-1}$  in irrigation water is recommended for the protection of agricultural crop species (CCME 1996). Data on the toxicity of cadmium to agricultural crops were available for 11 species from five plant families ranging from 1.8 to 16 mg \cdot kg^{-1} (Environment Canada 1994). Chronic data on four crop species within the grain family (Graminae) indicate that this group of plants is very sensitive to cadmium.

The acceptable soil concentration (ASC) is calculated as the geometric mean of the lowest quantifiable LOEL for crop plants ( $2.0 \text{ mg} \cdot \text{kg}^4$ ) resulting in a significantly reduced yield (20%) in rye (*Lolium perenne*) (Coppola et al. 1988) and the estimated corresponding NOEL of 0.44 mg·kg<sup>-1</sup> (NOEL = LOEL ÷ 4.5), and dividing by an uncertainty factor of 10 (CCME 1993). This ASC of 0.094 mg·kg<sup>4</sup> was then multiplied by the soil mass within 1 ha (soil bulk density 1300 kg·m<sup>3</sup>) × (soil bulk volume to a depth of 5 cm [i.e.,  $100 \times 100 \times 0.05$  m]) to calculate the allowable mass of cadmium in soil. A depth of 5 cm was chosen, since most of the cadmium (>95%) remains in surface soils for several years with only minor leaching occurring. The contaminant mass was then divided by the maximum irrigation rate ( $1.2 \times 10^7$  L·ha<sup>-1</sup> per year) to give the SMATC of 0.0051 mg·L<sup>4</sup> (i.e.,  $5.1 \mu$ g·L<sup>-1</sup>) in irrigation water (CCME 1996).

# **Livestock Water**

The recommended water quality guideline for cadmium for the protection of livestock is 80  $\mu$ g·L<sup>-1</sup> (CCME 1996). Data were available for five avian species, including two livestock species (Environment Canada 1994). Egg production, the most sensitive endpoint, was significantly reduced by 39% in domestic chickens after receiving a cadmium dose of 2.19 mg kg<sup>+</sup> bw per day (Leach et al. 1979). Similar reductions in egg production by domestic chickens were reported at cadmium doses of 2.28-2.4 mg·kg<sup>-1</sup> bw per day (Anke et al. 1970; Sell 1975). Data were available for 10 mammalian species, including four livestock species (Environment Canada 1994). The most sensitive livestock mammal was the sheep. Reduced body weight gain (20%) in sheep, the most sensitive endpoint, was reported after an oral cadmium dose of 1.87 mg·kg<sup>-1</sup> bw per day (Doyle et al. 1974). Similar reductions (30%) were reported for Holstein and Jersey cattle at a dose of 2.88 mg·kg<sup>4</sup> bw per day (Powell et al. 1964).

Table 1. Water quality guidelines for cadmium for the protection of agricultural water uses (CCME 1996).

Use	Guideline value ( $\mu g \cdot L^{4}$ )
Irrigation water	5.1*
Livestock water	80

<sup>\*</sup>Crop-specific based on sensitivity.

The TDI for each species is calculated as the geometric mean of the NOAEL and LOAEL divided by an uncertainty factor of 10. The most sensitive avian TDI was  $0.11 \text{ mg} \cdot \text{kg}^4$  per day based on the white leghorn chicken (NOAEL of 0.55 mg·kg<sup>-1</sup> bw per day, LOAEL of 2.19 mg·kg<sup>-1</sup> per day). The most sensitive mammalian TDI was 0.13 mg·kg<sup>-1</sup> per day, based on sheep (NOAEL of  $0.92 \text{ mg} \cdot \text{kg}^{-1}$  by per day, LOAEL of 1.87 mg \cdot \text{kg}^{-1} by per day). Multiplying these TDIs by the ratio of the animal body weight to water intake yielded the respective RCs. The lowest RC was calculated to be  $0.41 \text{ mg} \cdot \text{L}^{-1}$  from the TDI for the white leghorn chicken (body weight of 2.3 kg, water intake of  $0.61 \text{ L} \cdot \text{d}^{-1}$ ). To account for exposure to cadmium from sources other than water, the lowest RC is multiplied by an apportionment factor of 0.2, to give a water quality guideline of  $80 \,\mu g \cdot L^{-1}$  for the protection of livestock.

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